

FLOWABLE COMPOSITE

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Composite History

1956

Dr. R.L. Bowen

'bis-GMA'

1960's

"Adaptic"

1970's

Microfilled composite

Light-initiated composite

1980's

Posterior composite

1990's

Hybrid composite

Flowable, packable, etc.

2000's

Nanocomposite

Non-shrink composite ?



Development of Materials

USA

COMPOSITES



UK

GLASS-IONOMERS

BRITISH DEVELOPMENT

VISIBLE LIGHT CURING

AMERICAN DEVELOPMENT

RESIN-MODIFIED
GLASS-IONOMERS

Resin Composites & derived materials

■ Development

- Dissatisfaction with silicates/acrylics
- Development of 'Bowen's resin' – Bis-GMA
- Introduction of first materials – two paste systems
- Developments in filler content
 - Smaller particles
 - Microfine particles
- Command setting – UV cure
- Single paste – VLC
- Change in viscosity – flowable/packable
- Nanocomposites? Low shrinkage materials?

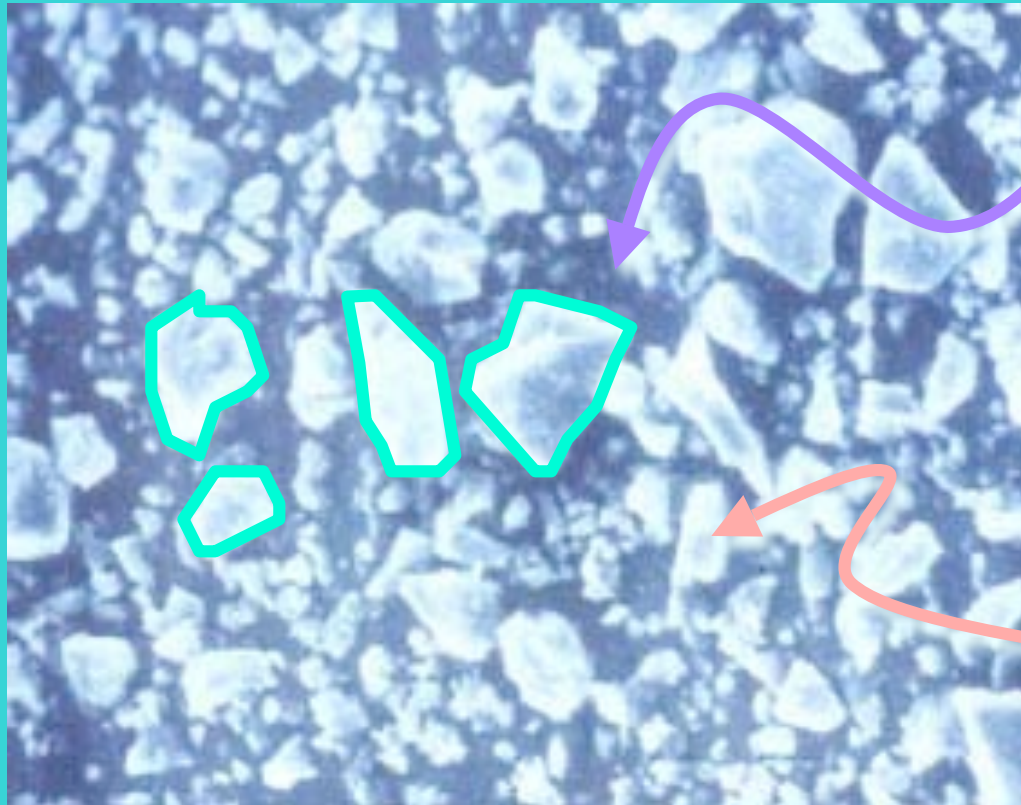
TECHNOLOGY

Alter the Paradigm



Review

Dental composite



Resin matrix

Ceramic fillers

Silane coupling
agent

Fillers

Variations

- Type

SiO_2 , barium glass, ZrO_2

- Size

macrofills ($>10 \mu\text{m}$); midifills ($1-10 \mu\text{m}$)
minifills ($0.1-1 \mu\text{m}$); microfills ($0.04-0.1 \mu\text{m}$);
nanofills ($0.02-0.07 \mu\text{m}$)

- Content

40% - 80⁺% by weight

Note : vol % is 15-20 % lower

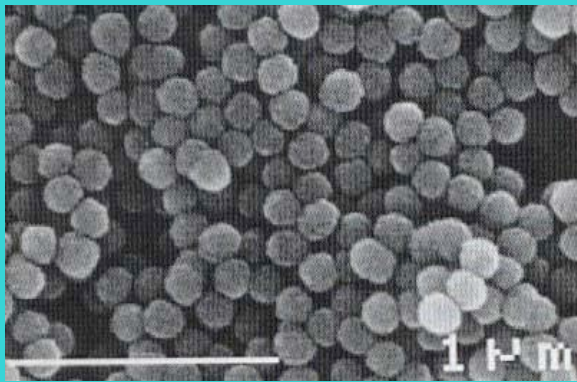
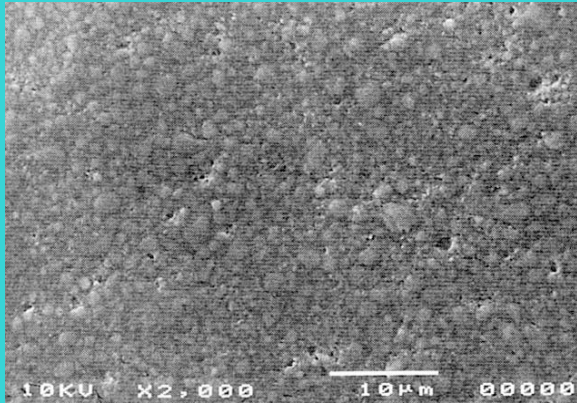
Classification

*Based on filler size
Lutz & Philips, 1983*

- **Traditional ('macrofilled')**
Glass particles ; size 1-15 μm
- **Microfilled**
Amorphous silica ; size $\sim 0.04 \mu\text{m}$
- **Hybrid**
Filler load $\sim 80 \text{ wt}\%$
Glass particles (Avg size $\sim 5 \mu\text{m}$)
+ microfillers

Fillers

Zirconia/silica



Z100, Z250, Palfique
(3M ESPE) (Tokuyama)

- sol-gel process
- spherical
- polishability
- continuum in sizes



high packing density
(85 wt %)

Resin matrix (light-activated composite)

Monomer : bis-GMA, bis-EMA, UDMA

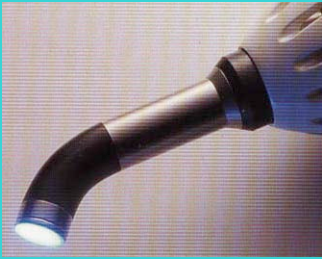
Diluent : TEGDMA

←
← Dimethacrylate

Photoinitiator : camphorquinone

Co-initiator : tertiary amine

Reaction

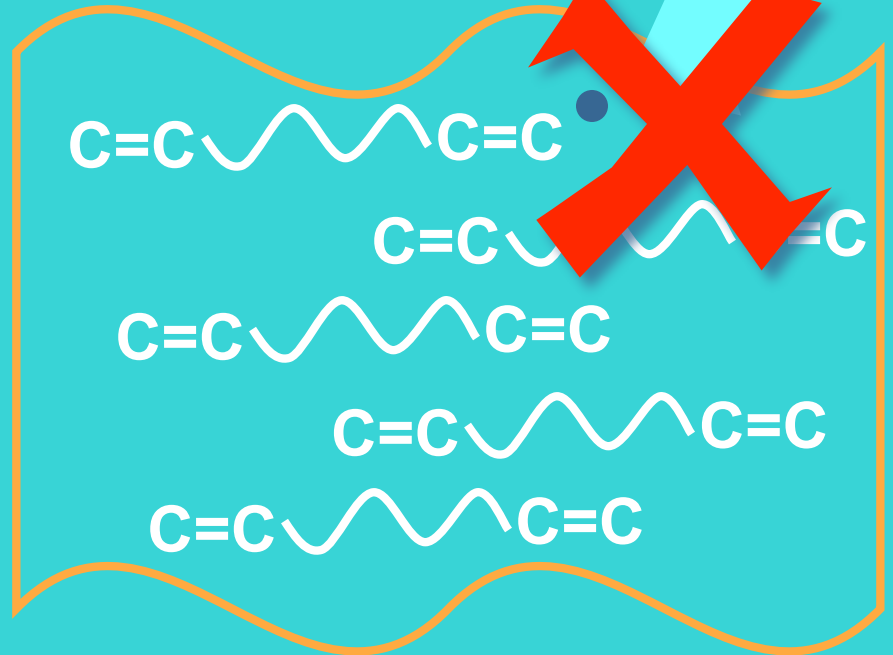
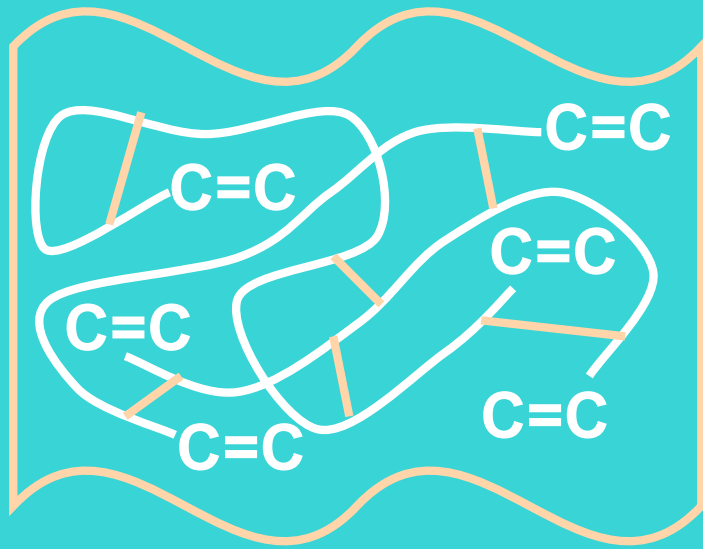


Amine co-initiator

CQ



+



Dimethacrylate monomer

Polymerization shrinkage

Current systems : 2-3 % vol shrinkage

- Methacrylate-based monomers
- Free radical, addition polymerization

Products in development : 'no shrinkage'

- Ring-opening reaction


expand

O₂ inhibition layer

= uncured monomer film at the surface

- film thickness ~ 10 μm
- not biocompatible → should be removed
- ? between increments ?

Biocompatibility

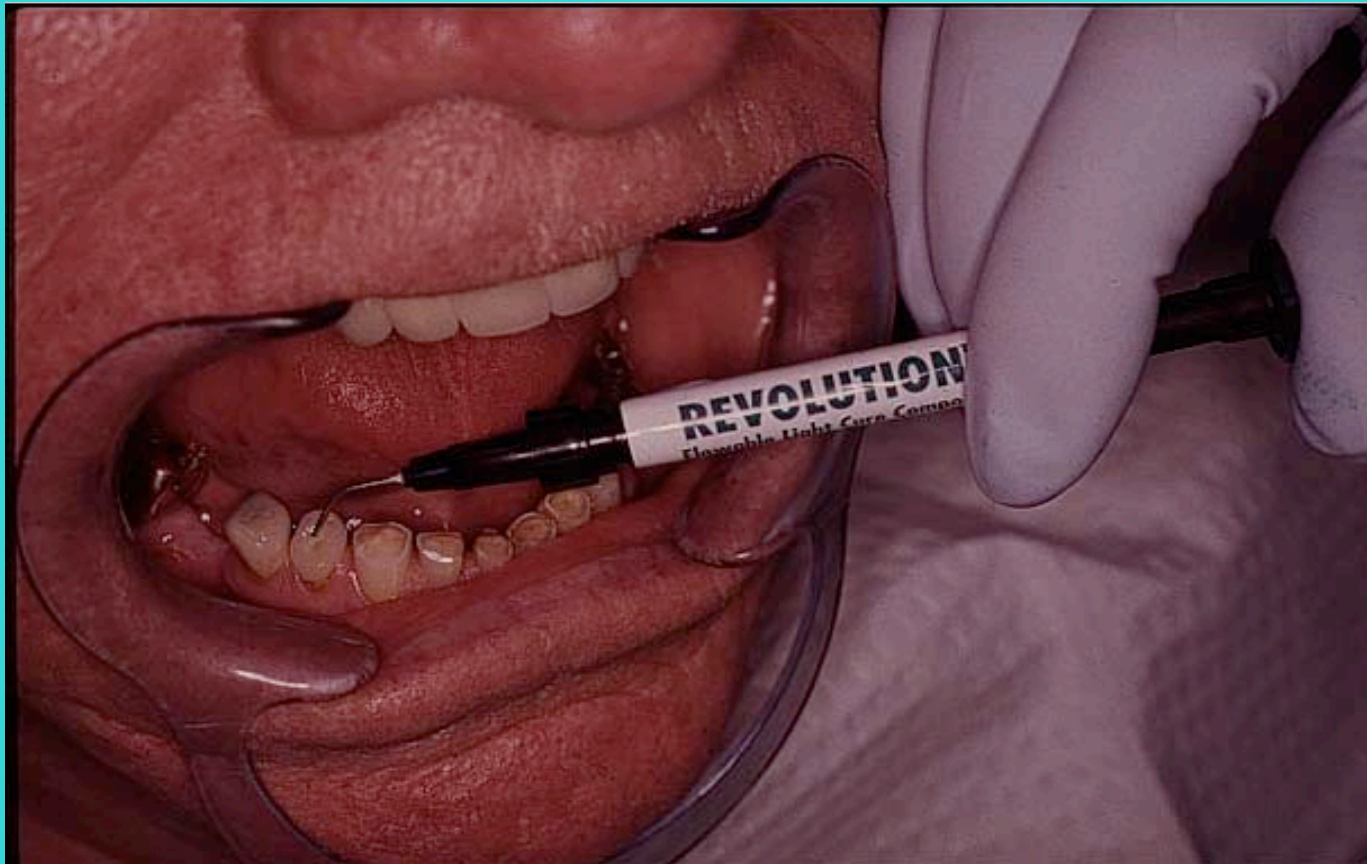
‘Most monomers have some biologic activity
.....but ... biocompatible when reacted
into polymer...’

Recommendations

- Well cured
- Remove O₂ inhibition layer
- Avoid contact with uncured resin

- 5% of dental personel have contact allergy to methacrylates
- Protective gloves are inadequate Wallenhammar et al, 2000
- Estrogenic effect Olea et al, 1996
- Adverse reaction: asthma, blister, rashes Hallstrom, 1993
- Oral lichenoid lesions Lind, 1998

FLOWABLE COMPOSITE



Problems with Paste Composite

- Difficult to use
- Difficult to manipulate
- Sticky, pull back
- Voids
- Porosities
- Unpolymerized areas
- Shrinkage
- Surface and Marginal integrity



Problems with Paste Composite

- Difficult to use
- Difficult to manipulate
- Sticky, pull back
- Voids
- Porosities
- Unpolymerized areas
- Shrinkage
- Surface and Marginal integrity
- Easy
- Easy
- Stays put
- Eliminates
- Eliminates
- Less
- Minimizes (technique)
- Best

Rheology



- *Definition*

- Rheology is the study of the flow and deformation of matter

- *Dental importance*

- Important for any material placed in the mouth in a fluid state – examples:
 - Impression materials
 - Directly-placed tooth restorative materials

Rheology of Resin Composites

VERY WIDE RANGE OF MATERIALS

SEALANTS

MICROFILL

FLOWABLE

HYBRID &

RELATED MATERIALS

PACKABLE or

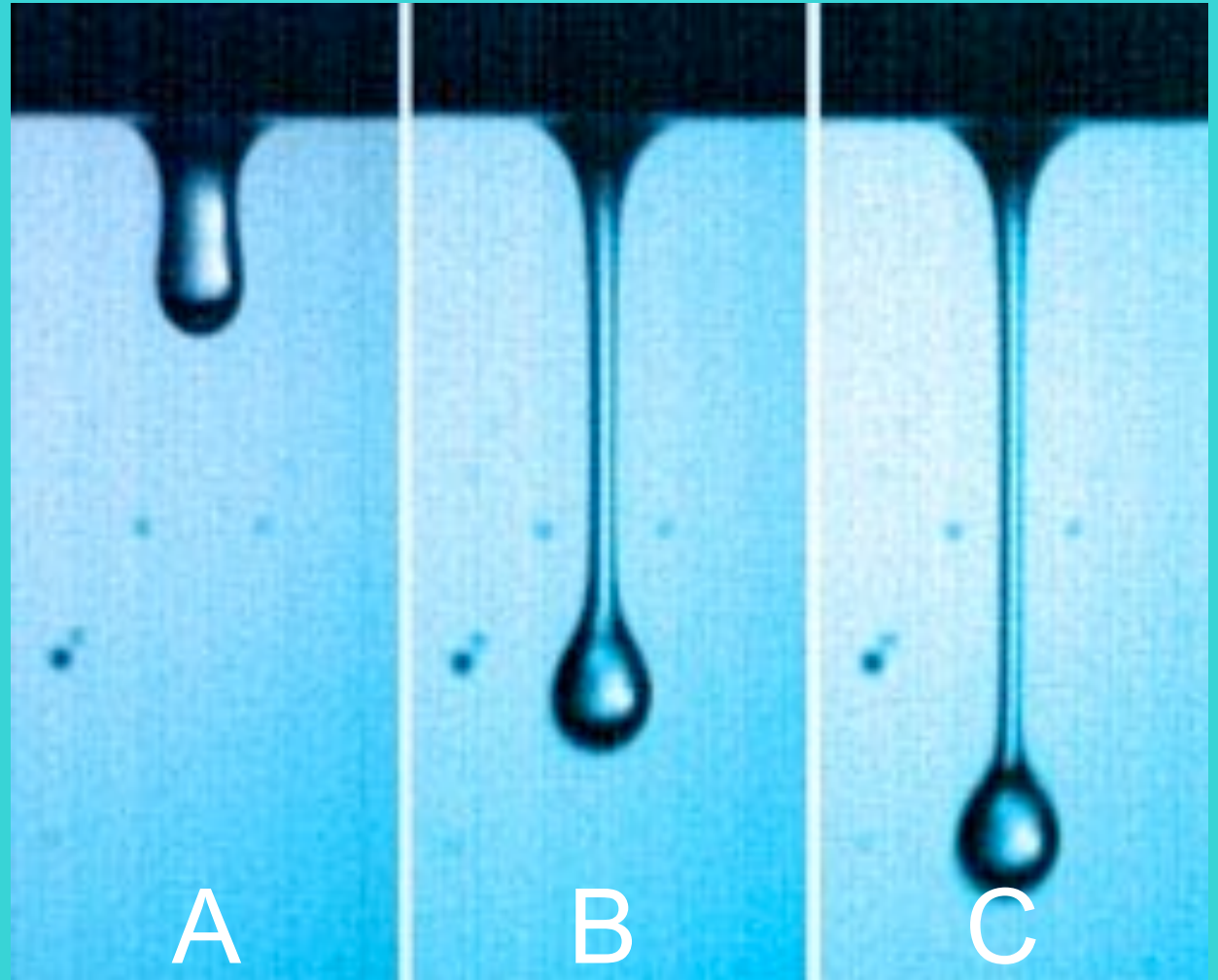
CONDENSABLE

INCREASING VISCOSITY



Viscosity

**A has highest
viscosity
C is the most
fluid
B is inter-
mediate**



**HIGH
FLUIDITY**

**LOW
VISCOSITY**



**LOW
FLUIDITY**

**HIGH
VISCOSITY**



Rheology of Resin Composites

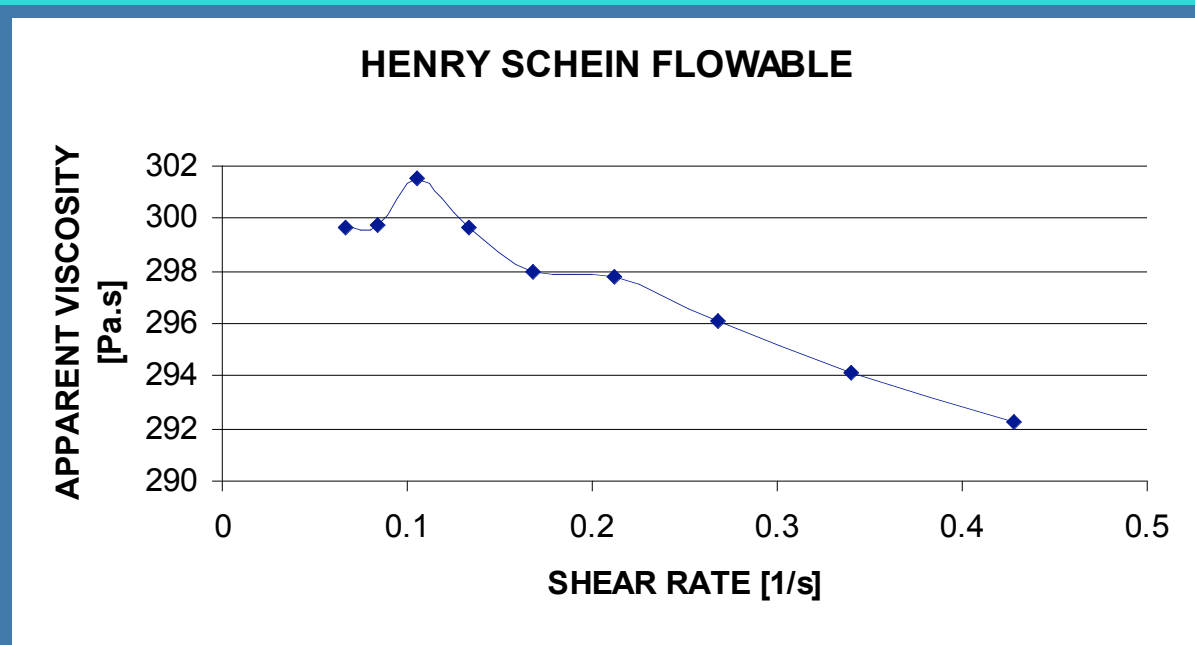
- General idea - lower filler content - usually hybrid filler. Includes fumed silica, claimed to give thixotropy*
- Original claims (patent literature) - recommended for Class I, II, III, IV and V cavities
- Now being recommended for many applications (e.g. fissure sealing – discussed later)
- *What is **thixotropy**?

Rheology of Resin Composites

- **Thixotropy** is a reversible structural breakdown of a material that occurs when the material is stressed
- **Flowable composites** were designed to be **thixotropic**
- This means that when the material is being syringed, the high stress from syringing breaks down some of the structure (e.g. hydrogen bonding), so the material flows
- But when the material is placed into the cavity, it will not flow ('non-drip') because the hydrogen bonding structure quickly recovers

Flowable Composites

THERE ARE *SIGNIFICANT* 'FLOW' PROPERTY DIFFERENCES BETWEEN DIFFERENT PRODUCTS

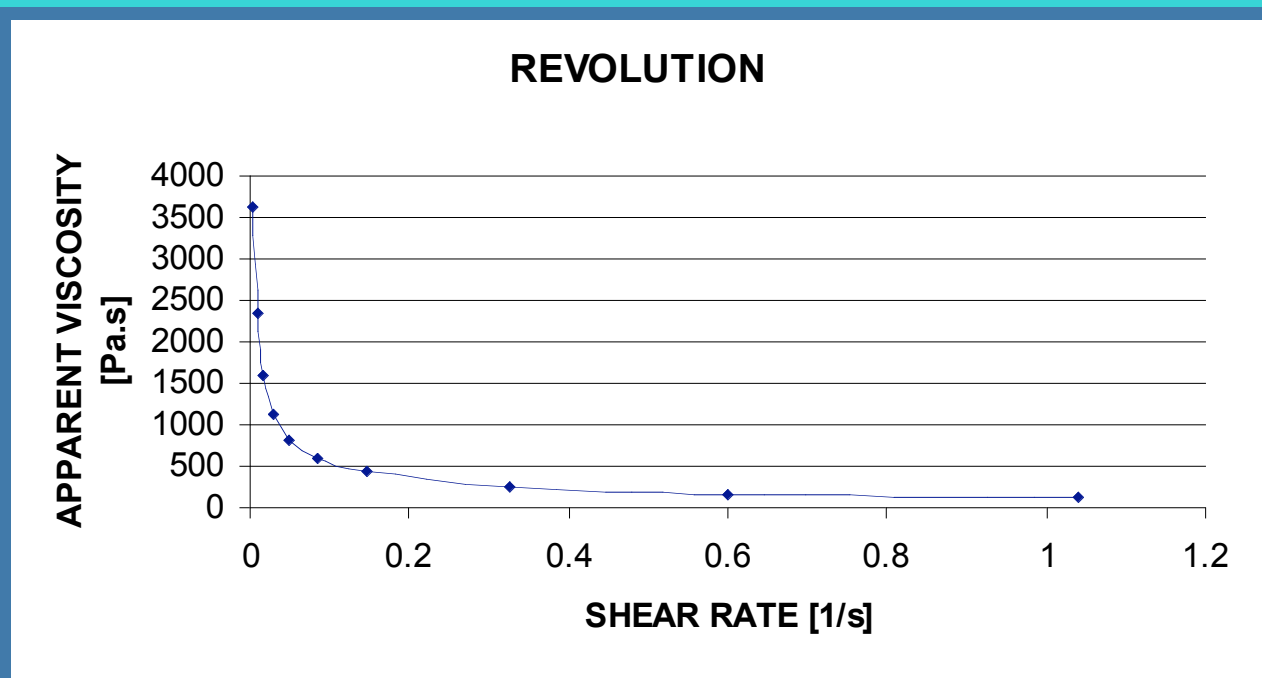


**VERY FLUID;
VIRTUALLY
NEWTONIAN:
NO
"THIXOTROPY"**

-- contrast with

Flowable Composites

THERE ARE *SIGNIFICANT* 'FLOW' PROPERTY DIFFERENCES BETWEEN DIFFERENT PRODUCTS



...contrast
HIGHER
VISCOSITY;
SHOWS
"SHEAR
THINNING"
DIFFERENT
TECHNIQUE
AND
APPLICATIONS?

Flowable Composites

THERE ARE *SIGNIFANT* COMPOSITIONAL DIFFERENCES BETWEEN DIFFERENT PRODUCTS

Material	Filler content (mass %)
<i>Henry Schein</i>	41
<i>Starflow</i>	61
<i>Revolution</i>	60
<i>Florestore</i>	50

Flowable Composites

- SOME MATERIALS (at least 341) – examples –
 - Florestore (Den-Mat)
 - Flowable composite (Henry Schein)
 - Revolution – Formula 2 (Kerr)
 - Starflow (Danville Materials)



Flowable Composites

- There is no such thing as a standard flowable material
- 34 brands (some identical); many with different formulations, handling characteristics, properties and applications
- Confusion because of untested materials - no track record of clinical success

1. Philosophy

- 1. Dental materials do not naturally belong in the mouth!
- 2. All synthetic materials evoke a host response
- 3. Synthetic materials not as good as health tooth substance
- 4. Best treatment is the least treatment
- 5. Best treatment is the most durable treatment

THEREFORE – there is an obligation to prevent dental disease, and where treatment is necessary, to choose the best materials, and manipulate them in such a way that optimum properties are obtained.

Principles of Selection

- (i) Evidence from laboratory data
- (ii) Clinical performance
- (iii) Esthetic considerations
- (iv) Clinical needs
- (v) Patient's preferences
- (vi) Operator's preferences
- (vii) Cost effectiveness
- (viii) Environmental considerations

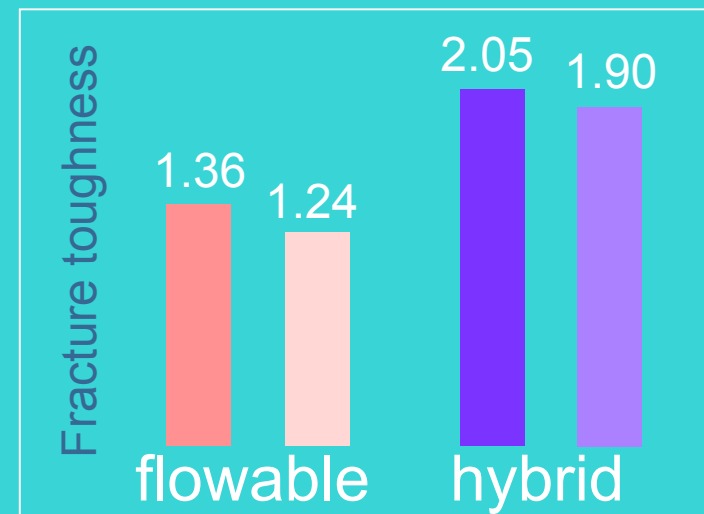
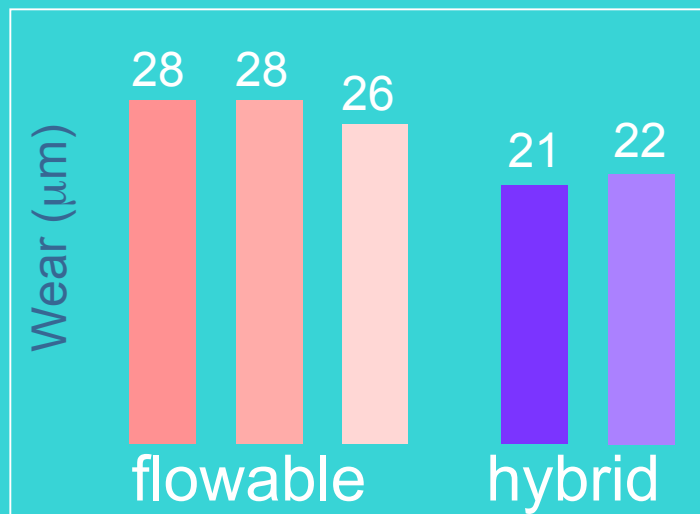
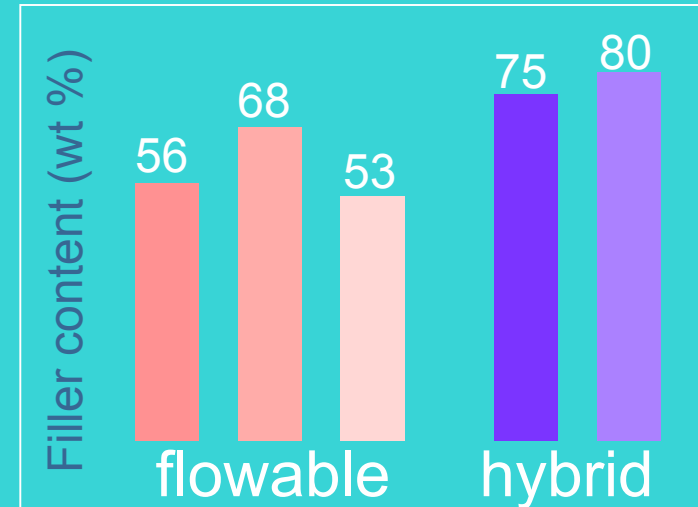
Flowable composites



Filler contents have
Caught up

] higher shrinkage

] compromised mechanical properties



Conclusion

- The criteria for choice of a material include not only factors such as physical and mechanical properties, but also include ease of manipulation.
- Note that there are no standard specifications for these materials, and that they differ considerably in their flow properties.

Need improvements?

B Biocompatibility cured vs uncured

I Interfacial properties adhesive system

C Chemical properties Biodegradation
esterase 

M Mechanical & physical properties

E Esthetic consideration & Polishability

P Practical questions Flowable / Packable

Failure Zone

- Biofilm
- Improper Etching
- Thick Layers
 - Unpolymerized areas
- Porosities
- Voids in the body
- Marginal Integrity
- Surface Integrity
- Shrinkage

FAILURE ZONE

- Biofilm
- Improper Etching
- Improper Curing
- Thick Layers
 - Unpolymerized areas
- Porosities
- Voids in the body
- Marginal Integrity
- Surface Integrity
- Shrinkage